

AMENDMENTS TO THE CLAIMS

Please amend the claims as follows:

1. (Currently amended) In a cell for electrowinning aluminium, the cell containing a molten electrolyte in which alumina is dissolved,

an An-anode that is in contact with the molten electrolyte for electrowinning aluminium from the alumina dissolved in a the molten electrolyte,

said anode comprising an electrically conductive substrate that is covered with an applied electrochemically active coating, said coating comprising a layer that contains predominantly cobalt oxide CoO.

2. (Original) The anode of claim 1, wherein the CoO-containing layer is a layer of sintered particles.

3. (Original) The anode of claim 1, wherein the CoO-containing layer is an integral oxide layer on an applied Co-containing metallic layer of the coating.

4. (Currently amended) The anode of ~~any preceding~~ claim 1, which comprises an oxygen barrier layer between the CoO-containing layer and the electrically conductive substrate.

5. (Original) The anode of claim 4, wherein the oxygen barrier layer contains at least one metal selected from nickel, copper, tungsten, molybdenum, tantalum, niobium and chromium, or an oxide thereof.

6. (Original) The anode of claim 5, wherein the oxygen barrier layer further contains cobalt.

7. (Original) The anode of claim 6, wherein the oxygen barrier layer is a cobalt alloy containing at least one metal selected from nickel, tungsten, molybdenum, tantalum and niobium.

8. (Original) The anode of claim 7, wherein the cobalt alloy contains:

- at least one of nickel, tungsten, molybdenum, tantalum and niobium in a total amount of 5 to 30 wt%, in particular 10-20 wt%; and
- one or more further elements and compounds in a total amount of up to 5 wt%, the balance being cobalt.

9. (Original) The anode of claim 8, containing as said further elements at least one of aluminium, silicon and manganese.

10. (Currently amended) The anode of ~~any one of claims~~ claim ~~4 to 9~~, wherein the CoO-containing layer is integral with the oxygen barrier layer.

11. (Currently amended) The anode of ~~any one of claims claim~~ claim 4 to 9, wherein the oxygen barrier layer is integral with the electrically conductive substrate.
12. (Currently amended) The anode of ~~any one of claims claim~~ claim 4 to 9, wherein the oxygen barrier layer and the CoO-containing layer, or precursors thereof, are distinct applied layers.
13. (Currently amended) The anode of ~~claim 3, or claim 11 or 12 when depending on claim 3~~, wherein the Co-containing metallic layer contains cobalt in an amount of at least 95 wt%, in particular more than 97 wt% or 99 wt%.
14. (Currently amended) The anode of ~~any one of claims claim~~ claim 3 to 13, wherein the Co-containing metallic layer contains at least one additive selected from silicon, manganese, nickel, niobium, tantalum and aluminium in a total amount of 0.1 to 2 wt%.
15. (Currently amended) The anode of ~~any preceding~~ claim 1, wherein the electrically conductive substrate comprises at least one metal selected from chromium, cobalt, hafnium, iron, nickel, copper, platinum, silicon, tungsten, molybdenum, tantalum, niobium, titanium, tungsten, vanadium, yttrium and zirconium, or a compound thereof, in particular an oxide, or a combination thereof.
16. (original) The anode of claim 15, wherein the electrically conductive substrate has an outer part made of cobalt or a cobalt-rich alloy to which the coating is applied.
17. (Currently amended) The anode of claim 16, wherein the outer part is made of a cobalt-rich alloy containing at least one of tungsten, molybdenum, tantalum and niobium, said cobalt alloy containing ~~in particular~~:
 - at least one of nickel, tungsten, molybdenum, tantalum and niobium in a total amount of 5 to 30 wt%, in particular 10-20 wt; and
 - one or more further elements and compounds in a total amount of up to 5 wt%,the balance being cobalt.
18. (Original) The anode of ~~any preceding~~ claim 1, wherein the electrically conductive substrate contains at least one oxidation-resistant metal, in particular a metal selected from nickel, cobalt, chromium and niobium.
19. (Original) The anode of claim 18, wherein the electrically conductive substrate consists essentially of at least one oxidation-resistant metal.
20. (Currently amended) The anode of ~~any preceding~~ claim 1, wherein the CoO-containing layer has an open porosity of up to 12%, in particular up to 7%.
21. (Currently amended) The anode of ~~any preceding~~ claim 1, wherein the CoO-containing layer has a porosity with an average pore size below 7 micron, in particular below 4 micron.

22. (Currently amended) The anode of ~~any preceding~~ claim 1, wherein the CoO-containing layer contains cobalt oxide CoO in an amount of at least 80 wt%, in particular more than 90 wt% or 95 wt%.

23. (Currently amended) The anode of ~~any preceding~~ claim 1, wherein the CoO-containing layer is substantially free of Co₂O₃ and substantially free of Co₃O₄.

24. (Currently amended) The anode of ~~any preceding~~ claim 1, wherein the CoO-containing layer is electrochemically active for the oxidation of oxygen ions and is uncovered or is covered with an electrolyte-pervious layer.

25. (Currently amended) The anode of ~~any one of claims~~ claim 1 to 23, wherein the CoO-containing layer is covered with an applied protective layer, in particular an applied oxide layer.

26. (Original) The anode of claim 25, wherein the applied protective layer contains cobalt oxide.

27. (Currently amended) The anode of claim 25 ~~or 26~~, wherein the applied protective layer contains iron oxide.

28. (Original) The anode of claim 27, wherein the applied protective layer contains oxides of cobalt and of iron, in particular cobalt ferrite.

29. (Currently amended) The anode of ~~any one of claims~~ claim 25 to 28, wherein the applied protective layer contains a cerium compound, in particular cerium oxyfluoride.

30. (Currently amended) The anode of ~~any one of claims~~ claim 25 to 29, wherein the applied protective layer is electrochemically active for the oxidation of oxygen ions and is uncovered or is covered with an electrolyte pervious-layer.

31. (Currently amended) The anode of ~~any preceding~~ claim 1, which has an electrochemically active surface that contains at least one dopant selected from iridium, palladium, platinum, rhodium, ruthenium, silicon, tungsten, molybdenum, tantalum, niobium, tin or zinc metals, Mischmetal, metals of the Lanthanide series, as metals and compounds, in particular oxides, and mixtures thereof.

32. (Original) The anode of claim 31, wherein the electrochemically active surface is made of an active material containing the dopant(s) in a total amount of 0.1 to 5 wt%, in particular 1 to 4 wt%.

33. (Currently amended) A method of manufacturing an anode as defined in ~~any preceding~~ claim 1, comprising:

- providing an electrically conductive anode substrate; ~~and~~
- forming an electrochemically active coating on the substrate by applying one or more layers onto the substrate, one of which contains predominantly cobalt oxide CoO; and
- contacting the anode with the molten electrolyte.

34. (Original) The method of claim 33, wherein the CoO-containing layer is formed by applying a layer of particulate CoO to the anode and sintering.
35. (Original) The method of claim 34, wherein the CoO-containing layer is applied as a slurry, in particular a colloidal and/or polymeric slurry, and then heat treated.
36. (Original) The method of claim 33, wherein the CoO-containing layer is formed by applying a Co-containing metallic layer to the anode and subjecting the applied metallic layer to an oxidation treatment to form said CoO-containing layer on said metallic layer, said CoO-containing layer being integral with said metallic layer.
37. (Original) The method of claim 36, wherein the oxidation treatment is carried out in an oxygen containing atmosphere, such as air.
38. (Currently amended) The method of claim 36 ~~or 37~~, wherein the oxidation treatment is carried out at a treatment temperature above 895°C or 920°C, preferably above 940°C, in particular within the range 950°C to 1050°C.
39. (Original) The method of claim 38, wherein the Co-containing metallic layer is heated from room temperature to said treatment temperature at a rate of at least 300°C/hour, in particular at least 450°C/hour, for example by being placed in an environment, in particular in an oven, that is preheated to said treatment temperature.
40. (Currently amended) The method of ~~claims claim~~ claim 37 ~~to 39~~, wherein the oxidation treatment at said treatment temperature is carried out for more than 8 or 12 hours, in particular from 16 to 48 hours.
41. (Currently amended) The method of ~~any one of claims claim~~ claim 35 ~~to 40~~, wherein the Co-containing metallic layer is further oxidised during use.
42. (Cancelled)
43. (Currently amended) The anode cell of claim 142, ~~wherein said anode is in contact with a molten electrolyte of the cell, in said cell wherein~~ the electrolyte is being at a temperature below 960°C, in particular in the range from 910° to 940°C.
44. (Currently amended) A method of electrowinning aluminium using an anode in a cell as defined in claim 142 ~~or 43~~, said method comprising passing an electrolysis current via the anode through the electrolyte to produce oxygen on the anode and aluminium cathodically by electrolysing the dissolved alumina contained in the electrolyte.
45. (Original) The method of claim 44, wherein oxygen ions are oxidised on the anode's CoO-containing layer.

Application No.: 10/591,636
Filed: September 5, 2006

46. (Currently amended) The method of claim 44 ~~or 45~~, wherein oxygen ions are oxidised on an active layer applied to the anode's CoO-containing layer that inhibits oxidation and/or corrosion of the anode's substrate.

47. (Cancelled)